# AUSTRALIAN CHEMICAL REAGENTS SAFETY DATA SHEET

Date Prepared: March 2021 Version No: 5

# 1. IDENTIFICATION OF THE MATERIAL AND SUPPLIER

I. IDENTIFICATI	ON OF THE MATERIAL AND SUPPLIER			
Product Name: Product Code: Other Names: Uses:	Multi Element Standard 1000 mg/L Cu Pb Zn Ni Co Fe Mn 2171 Nil Analytical Reagent			
Supplier:	Australian Chemical Reagents 38-50 Bedford Street Gillman SA 5013			
Contacts:	Telephone:       61 08 84402000         Fax:       61 08 84402001         Emergency Phone:       CHEMCALL 1800 127 406 - Australia +64-4-917-9888 -         International       +64-4-917-9888 -			
2. HAZARDS INF	FORMATION			
GHS Classifica				
Signal Word(s) Pictogram(s)	WARNING			
Hazard Stateme	ent (s) H290 May be corrosive to metals H315 Causes skin irritation. H319 Causes serious eye irritation. H335 May cause respiratory irritation.	H315 Causes skin irritation. H319 Causes serious eye irritation.		
Precautionary S Preventative	Statement(s) P234 Keep only in original container P264 Wash thoroughly after handling. P260 Do not breathe dust/fume/gas/mist/vapours/spray. P271 Use only outdoors or in a well-ventilated area. P280 Wear protective gloves/protective clothing/eye protection/face protection.			
Response	P302+P352 IF ON SKIN: Wash with plenty of soap and water. P312 Call a POISON CENTER or doctor/physician if you feel unwell. P363 Wash contaminated clothing before reuse. P304+P340 IF INHALED: Remove victim to fresh air and keep	at		

rest in a position comfortable for breathing.

P312 Call a POISON CENTER or doctor/physician if you feel unwell. P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. P337+P313 If eye irritation persists: Get medical advice/attention. P314 Get medical advice/attention if you feel unwell. P390 Absorb spillage to prevent material damage.

StorageP405 Store locked up.P403+P233 Store in a well-ventilated place. Keep container tightly closed.P406 Store in corrosive resistant/... container with a resistant inner liner.

# **Disposal** P501 Dispose of contents/container to an approved waste disposal plant.

# 3. COMPOSITION / INFORMATION ON INGREDIENTS

#### Ingredients :

Chemical Entity	CAS No	Proportion	
Copper chloride	[7447-39-4]	<1%	
Lead chloride	[7758-95-4]	<1%	
Zinc chloride	[7646-85-7]	<1%	
Nickel chloride	[7718-54-9]	<1%	
Cobalt chloride	[7646-79-9]	<1%	
Ferric chloride	[7705-08-0]	<1%	
Manganese chloride	[7773-01-5]	<1%	
Hydrochloric acid	[7647-01-0]	8%	
Water	[7732-18-5]	to 100%	

#### 4. FIRST AID MEASURES

Safety showers and eye wash facilities should be provided.

#### Swallowed :

If conscious wash out mouth with water. Seek medical advice. Show this SDS to medical practitioner. Eye :

Immediately hold eyelids open and flood with water for at least 15 minutes. Obtain medical aid. Show this SDS to medical practitioner.

#### Skin :

Remove contaminated clothing. Immediately wash skin thoroughly with water and mild soap. Seek medical advice if irritation persists. Show this SDS to medical practitioner. Launder clothing before reuse.

## Inhaled :

Remove from contaminated air. Maintain breathing with artificial respiration if necessary. Seek medical assistance. Show this SDS to a doctor.

# 5. FIRE FIGHTING MEASURES

## Suitable Extinguishing Media:

Water spray. Carbon dioxide, dry chemical powder, or appropriate foam.

# Hazards From Combustion Products:

Solutions will not burn or support combustion. Contact with metals such as zinc and tin will generate explosive hydrogen gas. Decomposition products include hydrogen chloride.

#### **Precautions For Fire Fighters and Special Protective Equipment:**

Fire fighters and others who may be exposed to combustion products during fire should wear full protective clothing including positive pressure self-contained breathing apparatus (SCBA). Wear SCBA with full face-piece, operated in positive pressure mode when fighting fires.

Hazchem Code: 2R

## 6. ACCIDENTAL RELEASE MEASURES

## Emergency procedures:

Prevent from entering waterways. Restrict access to area. Ventilate area. Remove chemicals that can react with the spilled material.

#### Methods and materials for containment and clean up:

Use inert material such as sand or earth to contain spill or leak. Neutralise with sodium bicarbonate. Absorb spills with chemical absorber or vermiculite and dispose of in accordance with local regulations.

# 7. HANDLING AND STORAGE

#### Precautions for Safe Handling:

Do not get in eyes, on skin, on clothing. Avoid prolonged or repeated exposure.

## Conditions for Safe Storage:

Store sealed in original container in a cool well ventilated situation away from foods and other chemicals. Do not store in direct sunlight. Observe good hygiene and housekeeping practices.

## 8. EXPOSURE CONTROLS / PERSONAL PROTECTION

#### National Exposure Standards:

Safe Work Australia – Hydrogen chloride 7.5mg/m<sup>3</sup> TWA & Peak Limitation Copper dusts and mists (as Cu) TWA 1mg/m<sup>3</sup> ;Lead, inorganic dusts or fumes (as Pb) TWA 0.15 mg/m<sup>3</sup> Nickel soluble compounds (as Ni) TWA 0.1mg/m<sup>3</sup> (Sensitiser) Cobalt, metal dusts or fumes (as Co) TWA 0.05 mg/m<sup>3</sup> Iron salts, soluble (as Fe) TWA 1mg/m<sup>3</sup>

#### Biological Limit Values: No data available.

#### **Engineering Controls:**

Not required with normal use. If mists are likely to be generated maintain atmospheric concentrations well below exposure standards with extraction ventilation.

#### **Personal Protective Equipment (PPE):**

The use of nitrile or neoprene gloves complying with AS 2161 and the use of faceshield, chemical goggles or safety glasses with side shield protection complying with AS/NZS 1337 is recommended.

#### 9. PHYSICAL AND CHEMICAL PROPERTIES

Appearance :
Odour:
pH:
Boiling Point ( <sup>0</sup> C) :
Freezing/melting Point:
Vapour Pressure (mm of Hg @ 25 <sup>o</sup> C):
Vapour Density:
Specific Gravity :
Flash Point ( <sup>0</sup> C) :
Flammability Limits (%) :
Solubility in Water (g/L) :

Nil 1 Not applicable Not applicable Not applicable 1.1 Not flammable Not flammable Soluble

Clear liquid

#### **10. STABILITY AND REACTIVITY**

Chemical stability:

Stable.

#### Conditions to avoid:

Acidic solution. Will corrode metals. Will produce toxic gases on contact with cyanides, sulphides, hypochlorites etc.

Incompatible materials:

Strong alkalies, powdered metals, hypochlorites, cyanides, sulphides.

# Hazardous decomposition products:

Refer to section 5 (Fire Fighting Measures).

#### Hazardous reactions:

Hazardous polymerization will not occur.

## 11. TOXICOLOGICAL INFORMATION

#### **Health Effects:**

Swallowed : Corrosive. For hydrochloric acid LD50 oral-rabbit 900mg/kg

**Eye :** Corrosive to eye tissue. 100mg rinse of hydrochloric acid produced mild irritation of rabbit eyes. **Skin :** Corrosive to skin.

**Inhaled :** Irritating to respiratory system. For hydrochloric acid LCLo inhalation-human 1300 ppm for 30 minutes

**Chronic Effects:** Nickel chloride is a cancer suspect agent. Repeated or prolonged skin contact may cause severe irritation or sensitisation

## 12. ECOLOGICAL INFORMATION

Ecotoxicity: No data available. Persistence and degradability: No data available. Mobility: No data available.

## **13. DISPOSAL CONSIDERATIONS**

Contact a licensed professional waste disposal service to dispose of this material. Observe all federal, state and local environmental regulations.

## 14. TRANSPORT INFORMATION

UN Number: 3264 UN Proper Shipping Name: CORROSIVE LIQUID ACIDIC INORGANIC N.O.S (Contains hydrochloric acid 20%) Class and subsidiary risk(s): 8 Packing Group: II Hazchem Code: 2R Special precautions for user : Nil

## **15. REGULATORY INFORMATION**

**Standard for the Uniform Scheduling of Drugs and Poisons (SUSDP):** Schedule 6

## **16. OTHER INFORMATION**

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END of SDS