

Infosafe No™ 3CHCB	Issue Date : February 2021	RE-ISSUED by ACR
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 Product Name **SULFURIC ACID 0.1-4.9%**

Classified as hazardous

## 1. Identification

<b>Chemical Product and Company Identification</b>	SULFURIC Acid 0.1 - 4.9% Manufacturer Address Chem-Supply Pty Ltd 38-50 Bedford St Gillman SA 5013
<b>GHS Product Identifier</b>	SULFURIC ACID 0.1-4.9%
<b>Company Name</b>	AUSTRALIAN CHEMICAL REAGENTS (ACR) (ABN 19 008 264 211)
<b>Address</b>	38 - 50 Bedford Street Gillman S.A. 5013 Australia
<b>Telephone/Fax Number</b>	Tel: (08) 8440 2000 Fax: (08) 8440 2001
<b>Emergency phone number</b>	CHEMCALL 1800 127 406 (Australia) / +64-4-917-9888 (International)
<b>Recommended use of the chemical and restrictions on use</b>	Laboratory reagent.

<b>Other Names</b>	<u><b>Name</b></u>	<u><b>Product Code</b></u>
	Sulphuric Acid 0.01N	0077
	Sulphuric Acid 0.02N	0078
	Sulphuric Acid 0.03N	0915
	Sulphuric Acid 0.04N	0822
	Sulphuric Acid 0.05N	5289
	Sulphuric Acid 0.125N	0080
	Sulphuric Acid 0.15N	2953
	Sulphuric Acid 0.16N	2774
	Sulphuric Acid 0.18N	4022
	Sulphuric Acid 0.1N	0079
	Sulphuric Acid 0.25N	3137
	Sulphuric Acid 0.2N	0081
	Sulphuric Acid 0.4N	3266
	Sulphuric Acid 0.5N	0082
	Sulphuric Acid 0.83%	5799
	Sulphuric acid 1% v/v	3528
	Sulphuric acid 1.25% w/w	5988
	Alkalinity Reagent (Taylor R-0009)	1212

**Other Information** EMERGENCY CONTACT NUMBER: +61 08 8440 2000  
 Business hours: 8:30am to 5:00pm, Monday to Friday.

Australian Chemical Reagents (ACR) does not warrant that this product is suitable for any use or purpose. The user must ascertain the suitability of the product before use or application intended purpose. Preliminary testing of the product before use or application is recommended. Any reliance or purported reliance upon Australian Chemical Reagents (ACR) with respect to any skill or judgement or advice in relation to the suitability of this product of any purpose is disclaimed. Except to the extent prohibited at law, any condition implied by any statute as to the merchantable quality of this product or fitness for any purpose is hereby excluded. This product is not sold by description. Where the provisions of Part V, Division 2 of the Trade Practices Act apply, the liability of Australian Chemical Reagents (ACR) is limited to the replacement of supply of equivalent goods or payment of the cost of replacing the goods or acquiring equivalent goods.

## 2. Hazard Identification

<b>GHS classification of the substance/mixture</b>	Corrosive to Metals: Category 1
<b>Signal Word (s)</b>	WARNING

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**Hazard Statement (s)** H290 May be corrosive to metals.

**Pictogram (s)** Corrosion



**Precautionary statement – Prevention** P234 Keep only in original container.

**Precautionary statement – Response** P390 Absorb spillage to prevent material damage.

**Precautionary statement – Storage** P406 Store in corrosive resistant container with a resistant inner liner.

**Precautionary statement – Disposal** P501 Dispose of contents/container to an approved waste disposal plant.

### 3. Composition/information on ingredients

Ingredients	Name	CAS	Proportion
	Water	7732-18-5	95-99.9 %
	Sulphuric acid	7664-93-9	0.1-4.9 %

### 4. First-aid measures

<b>Inhalation</b>	If inhaled, remove from contaminated area to fresh air immediately. Apply artificial respiration if not breathing. If breathing is difficult, give oxygen. Get medical aid if cough or other symptoms appear.
<b>Ingestion</b>	Rinse mouth thoroughly with water immediately, repeat until all traces of product have been removed. DO NOT INDUCE VOMITING. Seek medical advice if effects persist.
<b>Skin</b>	If skin or hair contact occurs, remove contaminated clothing and flush skin and hair with running water. Wash contaminated clothing before re-use. Seek medical advice.
<b>Eye contact</b>	If in eyes, hold eyelids apart and flush the eye continuously with running water. Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes. Seek medical advice if effects persist.
<b>First Aid Facilities</b>	Maintain eyewash fountain and normal washroom facilities in work area.
<b>Advice to Doctor</b>	Treat symptomatically based on judgement of doctor and individual reactions of the patient.
<b>Other Information</b>	For advice, contact a Poisons Information Centre (Phone eg Australia 13 1126; New Zealand 0800 764 766) or a doctor.

### 5. Fire-fighting measures

<b>Hazards from Combustion Products</b>	Irritating and highly toxic fumes and gases, including toxic oxides of sulfur (SO <sub>x</sub> ). Contact with most metals (such as aluminium, tin, lead and zinc) causes formation of flammable and explosive hydrogen gas (H <sub>2</sub> ). However, the risk is reduced due to the weaker concentration of sulfuric acid present.
<b>Specific Methods</b>	Use extinguishing media most appropriate for the surrounding fire. No limitations to the type of extinguishing media.
<b>Specific hazards arising from the chemical</b>	Material does not burn. Runoff may pollute waterways.
<b>Hazchem Code</b>	2R
<b>Precautions in connection with Fire</b>	Wear SCBA and structural firefighter's uniform.

### 6. Accidental release measures

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<b>Spills &amp; Disposal</b>	Neutralize with dilute sodium hydroxide, lime or sodium carbonate.
<b>Personal Precautions</b>	Avoid inhalation, contact with skin, eyes and clothing.
<b>Personal Protection</b>	Wear protective clothing specified for normal operations (see Section 8)
<b>Clean-up Methods - Small Spillages</b>	Absorb or contain liquid with sand, earth or spill control material. Shovel up using non sparking tools and place in a labelled, sealable container for subsequent safe disposal. Put leaking containers in a labelled drum or overdrum.
<b>Environmental Precautions</b>	Prevent from entering into drains, ditches, rivers or the sea.

## 7. Handling and storage

<b>Precautions for Safe Handling</b>	Avoid contact with eyes, skin, or clothing. May corrode metallic surfaces.
<b>Conditions for safe storage, including any incompatibilities</b>	Store in tightly closed containers, in a cool, dry, well-ventilated area away from incompatible substances.
<b>Corrosiveness</b>	Corrosive in presence of aluminium, zinc, stainless steel(304), stainless steel(316), copper. Moderate corrosive effect on bronze.
<b>Storage Regulations</b>	Refer Australian Standard AS 3780-2008 'The storage and handling of corrosive substances'.
<b>Storage Temperatures</b>	Store at room temperature (15 to 25 °C recommended).

## 8. Exposure controls/personal protection

Occupational exposure limit values	<u>Name</u>	STEL		TWA		<u>Footnote</u>
		<u>mg/m3</u>	<u>ppm</u>	<u>mg/m3</u>	<u>ppm</u>	
	Sulphuric acid	3		1		
<b>Other Exposure Information</b>	These Workplace Exposure Standards are guides to be used in the control of occupational health hazards. All atmospheric contamination should be kept to as low a level as is workable. These workplace exposure standards should not be used as fine dividing lines between safe and dangerous concentrations of chemicals. They are not a measure of relative toxicity. A time weighted average (TWA) has been established for Sulphuric acid (Safe Work Aust) of 1 mg/m <sup>3</sup> . The corresponding STEL level is 3 mg/m <sup>3</sup> . The STEL (Short Term Exposure Limit) is an exposure value that should not be exceeded for more than 15 minutes and should not be repeated for more than 4 times per day. There should be at least 60 minutes between successive exposures at the STEL. The exposure value at the TWA is the average airborne concentration of a particular substance when calculated over a normal 8 hour working day for a 5 day working week.					
<b>Appropriate engineering controls</b>	Maintain the concentrations values below the TWA. This may be achieved by process modification, use of local exhaust ventilation, capturing substances at the source, or other methods.					
<b>Respiratory Protection</b>	Where ventilation is not adequate, respiratory protection may be required. Avoid breathing vapours or mists. Select and use respirators in accordance with AS 1716 - Respiratory Protective Devices and be selected in accordance with AS 1715 - Selection, Use and Maintenance of Respiratory Protective Devices. When mists or vapours exceed the exposure standards then the use of the following is recommended: Approved respirator with organic vapour and dust/mist filters. Filter capacity and respirator type depends on exposure levels.					
<b>Eye Protection</b>	The use of a face shield, chemical goggles or safety glasses with side shield protection as appropriate. Must comply with Australian Standards AS 1337 and be selected and used in accordance with AS 1336.					
<b>Hand Protection</b>	Wear gloves of impervious material conforming to AS/NZS 2161: Occupational protective gloves - Selection, use and maintenance. Final choice of appropriate glove type will vary according to individual circumstances. This can include methods of handling, and engineering controls as determined by					

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	appropriate risk assessments. Avoid skin contact when removing gloves from hands, do not touch the gloves outer surface. Dispose of gloves as hazardous waste.
<b>Personal Protective Equipment</b>	Personal protective equipment should not solely be relied upon to control risk and should only be used when all other reasonably practicable control measures do not eliminate or sufficiently minimise risk. Guidance in selecting personal protective equipment can be obtained from Australian, Australian/New Zealand or other approved standards.
<b>Footwear</b>	Safety boots in industrial situations is advisory, foot protection should comply with AS 2210, Occupational protective footwear - Guide to selection, care and use.
<b>Body Protection</b>	Clean clothing or protective clothing should be worn, preferably with an apron. Clothing for protection against chemicals should comply with AS 3765 Clothing for Protection Against Hazardous Chemicals.
<b>Hygiene Measures</b>	Always wash hands before smoking, eating or using the toilet. Wash contaminated clothing and other protective equipment before storing or re-using.

## 9. Physical and chemical properties

<b>Form</b>	Liquid
<b>Appearance</b>	Clear, colourless liquid.
<b>Odour</b>	Odourless.
<b>Melting Point</b>	May start to solidify at -0.1 °C based on data for: water.
<b>Boiling Point</b>	~100°C
<b>Solubility in Water</b>	Miscible.
<b>Solubility in Organic Solvents</b>	Insoluble in methanol, diethyl ether, n-octanol (0.5%).
<b>Specific Gravity</b>	Approx. 1
<b>pH</b>	Acidic; pH of 0.01 N solution (~0.05%): 2.1; pH of 0.1 N solution (~0.5%): 1.2; pH of 1.0 N solution (~5.0%): 0.3.
<b>Flammability</b>	Non combustible material.
<b>Molecular Weight</b>	Sulfuric acid 98.08

## 10. Stability and reactivity

<b>Chemical Stability</b>	Stable under normal temperatures, pressures and conditions of use and storage.
<b>Conditions to Avoid</b>	Metals, excess heat, extremes of temperature, direct sunlight, combustible materials, organic materials, oxidizers, amines, bases, and incompatible materials.
<b>Incompatible Materials</b>	Alkali metals, alkaline earth metals, alkali compounds, ammonia, alkali hydroxide solutions, metals, metal alloys, organic solvents, permanganates.
<b>Hazardous Decomposition Products</b>	Irritating and highly toxic fumes and gases, including toxic oxides of sulfur (SO <sub>x</sub> ). Contact with most metals (such as aluminium, tin, lead and zinc) causes formation of flammable and explosive hydrogen gas (H <sub>2</sub> ). However, the risk is reduced due to the weaker concentration of sulfuric acid present.
<b>Possibility of hazardous reactions</b>	Flammable hydrogen gas is generated by the action of the acid on most metals (i.e. lead, copper, tin, zinc, aluminium, etc.). Reacts with alkali metals and alkaline earth metals.
<b>Hazardous Polymerization</b>	Will not occur.

## 11. Toxicological Information

<b>Toxicology Information</b>	No adverse health effects expected if the product is handled in accordance with this Safety Data Sheet and the product label. If mishandled or overexposed to this product the following symptoms or effects may occur.
<b>Ingestion</b>	Ingestion of this product may cause irritation and possible burns of mucous membranes in the mouth, pharynx, oesophagus, and gastrointestinal tract,

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<b>Inhalation</b>	causing nausea, vomiting and diarrhoea. Inhalation of product vapours may cause irritation to the mucous membranes of the nose, throat and respiratory system, with sore throat, coughing, and shortness of breath.
<b>Skin</b>	May causes irritation to skin and mucous membranes. Symptoms may include redness, itching, and pain.
<b>Eye</b>	Direct contact with eyes may cause temporary irritation. Symptoms may include tearing, blurred vision, redness, stinging, and pain.
<b>Respiratory sensitisation</b>	Not classified based on available information.
<b>Skin Sensitisation</b>	Not classified based on available information.
<b>Germ cell mutagenicity</b>	Not classified based on available information.
<b>Carcinogenicity</b>	Not classified based on available information.
<b>Reproductive Toxicity</b>	Not classified based on available information.
<b>STOT-single exposure</b>	Not classified based on available information.
<b>STOT-repeated exposure</b>	Not classified based on available information.
<b>Serious eye damage/irritation</b>	Not classified based on available information.
<b>Mutagenicity</b>	Not classified based on available information.
<b>Skin corrosion/irritation</b>	Not classified based on available information.

## 12. Ecological information

<b>Ecological Information</b>	No ecological problems are to be expected when the product is handled and used with due care and attention.
<b>Ecotoxicity</b>	Harmful effect due to pH shift. Quantitative data on the ecological effect of this product are not available. The following applies to sulfuric acid in general: Harmful effect on aquatic organisms. Toxic effect on fish and algae. Caustic even in diluted form. Does not cause biological oxygen deficit. Endangers drinking-water supplies if allowed to enter soil and/or waters in large quantities. Neutralisation possible in waste water treatment plants.
<b>Bioaccumulative Potential</b>	An enrichment in organisms should not be expected.
<b>Short Summary of Assessment of Environmental Impact</b>	When released into the soil, this material may leach into groundwater. When released into the air, this material may be removed form the atmosphere to a moderate extent by wet and dry deposition.
<b>Environmental Protection</b>	Do not allow to enter waters, waste water, or soil!
<b>Acute Toxicity - Daphnia</b>	Daphnia magna EC50: 29 mg/l /24 h (pure substance).

## 13. Disposal considerations

<b>Disposal Considerations</b>	Dispose of according to relevant local, state and federal government regulations.
<b>Waste Disposal</b>	Neutralise remaining product with lime, soda ash or sodium bicarbonate, adjusting pH to 6-8. Flush to sewer as greatly diluted solution.

## 14. Transport information

<b>Transport Information</b>	Dangerous Goods of Class 8 Corrosives are incompatible in a placard load with any of the following: - Class 1, Class 4.3, Class 5, Class 6, if the Class 6 dangerous goods are cyanides and the Class 8 dangerous goods are alkalis and
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<b>U.N. Number</b>	Class 7. 2796
<b>UN proper shipping name</b>	SULFURIC ACID
<b>Transport hazard class(es)</b>	8
<b>Hazchem Code</b>	2R
<b>Packing Group</b>	II
<b>EPG Number</b>	8A1
<b>IERG Number</b>	37
<b>Environmental Hazards</b>	Harmful effect due to pH shift. The following applies to sulphuric acid: Harmful effect on aquatic organisms. Toxic effect on fish and algae. Neutralisation possible in waste water treatment.
<b>Other Information</b>	There is a possibility that this product could be contained in a reagent set or kit composed of various compatible dangerous goods. If the item is not in a reagent set or kit, the classification given above applies. If the item is part of a reagent set or kit the classification would change to the following: UN3316 Chemical Kit, Hazard Class 9, Packing Group II or III. If the item is not regulated, the Chemical Kit classification does not apply.

## 15. Regulatory information

<b>Regulatory Information</b>	Listed in the Australian Inventory of Chemical Substances (AICS). Not listed under WHS Regulation 2011, Schedule 10 - Prohibited carcinogens, restricted carcinogens and restricted hazardous chemicals.
<b>Poisons Schedule</b>	S6

## 16. Other Information

<b>Literature References</b>	'Standard for the Uniform Scheduling of Medicines and Poisons .', Commonwealth of Australia. National Road Transport Commission, 'Australian Code for the Transport of Dangerous Goods by Road and Rail 7th. Ed.' Safe Work Australia, 'National Code of Practice for the Preparation of Safety Data Sheets for Hazardous Chemicals'. Standards Australia, 'SAA/SNZ HB 76:2010 Dangerous Goods - Initial Emergency Response Guide', Standards Australia/Standards New Zealand. Safe Work Australia, 'Hazardous Chemical Information System'. Safe Work Australia, 'National Code of Practice for the Labelling of Safe Work Hazardous Substances'. Safe Work Australia, 'National Exposure Standards for Atmospheric Contaminants in the Occupational Environment'.
<b>Contact Person/Point</b>	Paul McCarthy Ph. (08) 8440 2000 <b>DISCLAIMER STATEMENT:</b> All information provided in this data sheet or by our technical representatives is compiled from the best knowledge available to us. However, since data, safety standards and government regulations are subject to change and the conditions of handling and use, or misuse, are beyond our control, we make no warranty either expressed or implied, with respect to the completeness or accuracy to the information contained herein. Australian Chemical Reagents (ACR) accepts no responsibility whatsoever for its accuracy or for any results that may be obtained by customers from using the data and disclaims all liability for reliance on information provided in this data sheet or by our technical representatives.
<b>Empirical Formula &amp; Structural Formula</b>	H2SO4 + aqua  ...End Of MSDS...

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